

# Size-mobility relation in the nanometer range

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## 1. The Millikan model

mean free path of gas molecules

the slip factor coefficients  
( $a = 1.2$ ,  $b = 0.5$ ,  $c = 1$ )

$$K = e \frac{1 + \frac{l}{\delta} \left[ a + b \exp\left(-c \frac{\delta}{l}\right) \right]}{6\pi\eta\delta}$$

gas viscosity

collision distance

## 2. The Chapman-Enskog model

elementary charge

mass of gas molecule

mass of ion

$$K_{\text{free molecule}} = \frac{3e}{8n_g \Omega^{(1,1)}} \sqrt{\frac{\pi(1 + m_g / m_i)}{2m_g kT}}$$

number concentration of molecules

first collision integral

Boltzmann constant

absolute temperature

Charged rigid spheres and ( $\infty-4$ ) potential:

( $\delta$  = the collision distance).

$$\Omega^{(1,1)} = \Omega^{(1,1)*} \pi \delta^2, \quad \Omega^{(1,1)*} = f(T^*), \quad T^* = \frac{kT}{U(\delta)}, \quad U(\delta) = \begin{cases} \text{if } r < \delta \text{ then } \infty \\ \text{if } r > \delta \text{ then } \frac{-\alpha e^2}{8\pi\epsilon_0 r^4} \end{cases}$$

$$f(T^*) = \begin{cases} \text{if } T^* \leq 1 \text{ then } 1.4691 \times T^{*-1/2} - 0.341 \times T^{*-1/4} + 0.185 \times T^{*5/4} + 0.059 \\ \text{if } T^* \geq 1 \text{ then } 1 + 0.106 \times T^{*-1} + 0.263 \times T^{*-4/3} \end{cases}$$

Size ???

### 3. New model (modified Millikan equation)

$$K = f_1 f_2 e^{\frac{1 + \frac{l}{\delta} \left[ a + b \exp\left(-c \frac{\delta}{l}\right) \right]}{6\pi\eta\delta}}$$

$$\delta = r_i + h_{(T_\delta)} + r_g(T_\delta),$$

$h$  – extra distance

$$f_1 = \sqrt{1 + \frac{m_g}{m_i}},$$

$$f_2 = \frac{2.25}{(a + b) \left( \Omega_{\infty-4}^{(1,1)*}(T^*) + s(r_i, T_\delta) - 1 \right)}$$

When  $r_m \rightarrow 0$  the model approaches the free molecule regime equation with

$$\Omega^{(1,1)} = \left[ \Omega_{\infty-4}^{(1,1)*} + s(r_i, T_\delta) - 1 \right] \pi \delta^2$$

### 4. Factor $s(r_i, T_\delta)$

*Assumptions:*

- 1) The melting of internal degrees of freedom of the particle energy is described by the Einstein factor:

$$\frac{x^2 e^x}{(e^x - 1)^2} \quad \text{where} \quad x = \frac{\Delta E}{kT}$$

- 2) the average separation of internal energy levels is inversely proportional to the number of atoms  $\Delta E = \frac{\text{const}}{r_i^3}$

It follows the model  $s = 1 + (s_\infty - 1)x^2 \frac{e^x}{(e^x - 1)^2}$  where  $x = \frac{273 \text{ K}}{T} \left( \frac{r_{cr}}{r_i} \right)^3$ .

$r_{cr}$  – critical radius.